Optical Pumping in He³[†]

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Nuclear polarization of He³ has been performed using optical pumping techniques. The polarization is produced in the triplet metastable levels and transferred to the ground-state atoms by metastability exchange processes. The cross section for this process has been measured by two different methods, giving the results σ = (6.1 \pm 1) \times 10⁻¹⁶ cm² and σ = (7.9 \pm 2) \times 10⁻¹⁶ cm². One of the methods involved a spectroscopic determination of the density of metastable atoms in a weak discharge in He³, which was found to vary between 12.5 and $3.5\times10^{10}/\text{cc}$. It was shown that one hyperfine component only $(F = \frac{1}{2} \rightarrow \frac{3}{2})$ of the $2^{3}S_{1} - 2^{3}P_{0}$ line of He³ is excited by light from a He⁴ lamp. For this reason a previously quoted relation between the degree of polarization and the optical pumping signal needs to be modified.

 $\rm A$ RECENT paper by Colegrove, Walters, and Schearer¹ describes a He 3 optical pumping experi-RECENT paper by Colegrove, Walters, and ment in which polarization produced in the $2^{3}S_1$ metastable levels of He³ by optical pumping is transferred to the ${}^{1}S_{0}$ ground level by a process of metastability exchange. We have performed similar experiments but in view of the completeness of the work of Colegrove *et al.,* only additional points of interest will be reported here.

The cross section for metastability exchange between ground-state and metastable atoms was determined from observations of the transverse relaxation of ground-state atoms, together with a determination of the density of metastable atoms. The result of applying a strong rotating field $H₁$, with frequency equal to the ground-state Larmor frequency γH_0 , is to cause nutation of the macroscopic ground-state magnetization, with the frequency γH_1 . For all normal field strengths H_1 , the period $1/\gamma H_1$ is much greater than the time taken for the metastable atoms to exchange metastability with the ground-state atoms. This means that the nutation can be detected optically by monitoring the absorption in the metastable levels. The nutational transient so observed decays with time constant θ where

$2/\theta = 1/T_r + 1/T_2$.

Tr is the relaxation time of the longitudinal magnetization of atoms in the ground state in the absence of the rotating field, and T_2 is the transverse relaxation time under the conditions of the experiment. *T^r* can be measured independently (see Ref. 1), and is found to be long (>100 sec) compared with $T_2(\sim 0.1 \text{ sec})$. The transverse relaxation is almost entirely due to metastability exchange processes, if metastable atoms are present. We therefoie obtain as a good approximation the relation

$$
2/\theta\!\approx\!1/T_2\!=\!\sigma v n\,,
$$

where σ is the cross section for metastability exchange,

v is the mean relative velocity of the He atoms at the temperature of the cell, and *n* is the density of metastable atoms. T_2 was determined for a number of values of *n,* from photographs of the pumping transient obtained for different discharge conditions.

The quantity *n* was measured using a scanning Fabry-Perot interferometer² to determine the absorption

FIG. 1. Spectral profile of the $2^{3}S_{1}-2^{3}P_{1,2}$ fine-structure components from He⁴ lamp, as seen through He³ absorption cell. In the upper half of the figure the upper trace shows the profile with the discharge off in the absorption cell, for the lower trace the discharge was on. The lower half of the figure shows the absorption coefficient calculated from the above traces. The instrumental width was estimated to be 0.03 cm-1 . When "folded" with a Doppler width of 0.065 cm⁻¹ as expected for He³ at 290° K, we obtain a width of 0.085 cm^{-1} which is very close to that actually observed (0.08 cm⁻¹). The absorption peak is due to the $F = \frac{1}{2} \rightarrow \frac{3}{2}$ hyperfine component of the $2^{3}S_{1} - 2^{3}P_{0}$ line of He³. The tail on the left is the beginning of the peak due to the $F = \frac{1}{2} \rightarrow \frac{1}{2}$ hyperfine component.

2 H. G. Kuhn and H. J. Lucas-Tooth, J. Sci. Instr. 35, 413 (1953).

f This work was supported by the Rutherford Institute (Harwell, England).

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fornia, Berkeley, California. 1 F. D. Colegrove, L. D. Schearer, and G. K. Walters, Phys. Rev. **132,** 2561 (1963).

profile of the metastable He³ atoms, under the discharge conditions used in the experiment. The density is related to the absorption coefficient *kv* by the relation

$$
\int_0^\infty k_\nu d\nu = \frac{\pi e^2}{mc^2} nf,
$$

where f is the oscillator strength of the transition.

Ideally a white light source would be used for absorption experiments, but intensity considerations led us to try a He³ lamp, in the hope that the emission lines from the hot lamp would be sufficiently Doppler broadened to allow a good absorption profile to be obtained. In practice, accurate values of the absorption coefficient could be obtained in the center of the line only by this method, the lamp having insufficient intensity in the wings of the lines, where absorption was weak, to allow accurate measurements. The difficulty was overcome by using a He⁴ lamp as the source, as in the optical pumping experiments. The unresolved He⁴ component $D_3(2^3S_1-2^3P_{1,2})$ was sufficiently broad to overlap completely the $F = \frac{1}{2} \rightarrow \frac{3}{2}$ hyperfine component of the $\text{He}^{3}(2^{3}S_{1}-2^{3}P_{0})$ line. A sample trace of the absorption profile is given in Fig. 1 together with the calculated absorption coefficient. The width of this profile is in agreement with that predicted from considerations of Doppler and instrumental widths.

The value of f was taken to be 0.58,³ and the metastable density was found to vary between 12.5 and $3.5\times10^{10}/c$ c, according to discharge intensity, in a cell filled to 1-mm Hg pressure. The absolute accuracy of these figures is probably not better than 25% , owing mainly to the ill-defined absorption length in a spherical cell, and the fact that the discharge was somewhat inhomogeneous. A cylindrical cell was tried, but the

discharge became even more inhomogeneous, so that little advantage was gained. Relative measurements of the density for different discharge conditions were evidently better than this, as a plot of $1/T_2$ against metastable density shows reasonable linearity (Fig. 2). As a result of these measurements we found $\sigma = (7.9 \pm 2)$ $\times 10^{-16}$ cm².

A second determination of σ was made using the method described in Ref. 1, which relies on a measuiement of the linewidth of the metastable-atom resonances. If the characteristic time between exchange collisions of metastable atoms with ground-state atoms is τ_2 then

$$
1/\tau_2 = \sigma v N,
$$

where *N* is the density of ground-state atoms. This method is subject to an uncertainty in *N,* which is determined by the pressure in the cell. This is a rather doubtful quantity for a sealed-off absorption cell. By this method we obtained the following result:

$$
\sigma = (6.1 \pm 1) 10^{-6} \text{ cm}^2.
$$

This may be compared with the result of Colegrove *et al.*:

$$
\sigma = (4.2 \pm 0.5) 10^{-16} \,\mathrm{cm}^2.
$$

The experiments on the absorption of He⁴ light by He³ atoms were useful in that they verified part of the work of Fred *et al⁴* on the isotope shift and hyperfine struc-

10 830 A LINE STRUCTURE IN He³ AND He⁴

FIG. 3. The theoretical predictions of Fred *et al.* for the $2 \, \mathrm{{}^{8}S_1}$ -2 ³ $P_{0,1,2}$ 10 830 Å transitions of He³ and He⁴, compared with our lamp profiles. In each case the relative positions of He³ and He⁴ are drawn correctly.

⁴M. Fred, F. S. Tomkins, J. K. Brody, and M. Hammermesh, Phys. Rev. 82, 406 (1951).

³ C. W. Allen, *Astro physical Quantities* (Athlone Press, 1963).

ture of He³. They also brought to light a slight discrepancy, concerning Fig. 1 of their paper, which is also reproduced in Ref. 1. The figure is incorrectly drawn, so that the He³ and He⁴ profiles overlap more than they should. The reason is that different scales are used in the two parts of the figure. Figure 3 shows the theoretical predictions of Fred *et ah,* correctly drawn, and below are shown the profiles of our He³ and He⁴ lamps also in their correct relative positions. The importance of this is in relating the optical pumping signal observed in the experiments to the polarization of the sample. Colgrove *et al.* assumed that both hyperfine components of the He³ $2^{3}S_1 - 2^{3}P_0$ line were equally absorbent to He⁴ light, and obtained the relation:

$$
\frac{\Delta I}{I} = \frac{P(11 - 2P - P^2)}{(6 + 2P^2)} \frac{11P}{6}, \text{ if } P \text{ is small.}
$$

In view of the above, this should be amended to

$$
\frac{\Delta I}{I} = \frac{P(15-10P+3P^2)}{(6+2P^2)} \frac{5P}{2}
$$
, if P is small.

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^-Ionization Cross Sections for Relativistic Electrons

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Experimental values of the X-shell ionization cross section are determined from scintillation spectrometer measurements of the *K* x rays emitted when thin targets of tin and gold are bombarded by 50-, 100-, 200-, and 500-keV electrons. For these energies and atomic numbers, the experimental results show differences from the cross sections given by Burhop's nonrelativisitic calculations and by Perlman's relativistic calculations, but show good agreement with the predictions of Arthurs and Moiseiwitsch, who employed relativistic free-particle wave functions for the projectile electron and nonrelativistic wave functions for the atomic electron.

1. INTRODUCTION

THEOERTICAL studies of atomic K-shell ioniza-
tion by electrons may be divided into two main
groups. First are the nonrelativistic calculations by HEOERTICAL studies of atomic K-shell ionization by electrons may be divided into two main

FIG. **1.** Dependence of the parameter *S* on the ratio of the initial electron kinetic energy T_1 to the K-shell binding energy I_K . The parameter S is obtained from the calculations of Arthurs and Moiseiwitsch (Ref. 7) and is related to the K-ionization cross section by Eq. (i) .

Bethe,¹ Massey and Mohr,² Soden,³ Massey and Burhop,⁴ Wetzel,⁵ and Burhop,⁶ and second are the relativistic calculations by Arthurs and Moiseiwitsch⁷ and Perlman.⁸ In most cases, these calculations are not in a simple analytical form and must be evaluated numerically. Burhop⁶ has evaluated the K-ionization cross sections for the higher atomic numbers, 28 (nickel), 47 (silver), and 80 (mercury), and among the nonrelativistic calculations listed above, his results are most pertinent to the present investigation. Arthurs and Moiseiwitsch have calculated values for a parameter *S* from which the *K*-ionization cross section σ can be obtained for electron kinetic energies T_1 extending

¹ H. Bethe, Ann. Phys. 5, 325 (1930).
² H. S. W. Massey and C. B. O. Mohr, Proc. Roy. Soc. (London)
A140, 613 (1933). 3 D. Graf Soden, Ann. Physik 19, 409 (1934).

- ⁴H. S. W. Massey and E. H. S. Burhop, Phys. Rev. 48, 468 (1935)
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- 5 W. Wetzel, Phys. Rev. 44, 25 (1933). ⁶E. H. S. Burhop, Proc. Cambridge Phil. Soc. 36, 43 (1940). 7 A. M. Arthurs and B. L. Moiseiwitsch, Proc. Roy. Soc. (London) A247, 550 (1958).
- ⁸H. S. Perlman, Proc. Phys. Soc. (London) **76,** 623 (1960).